

Enantioselective Reformatsky Reaction with Ketones. Asymmetric Synthesis of β -(*tert*-Hydroxy)esters

Kenso Soai,* Atsushi Oshio and Takaki Saito

Department of Applied Chemistry, Faculty of Science, Science University of Tokyo, Shinjuku, Tokyo, 162 Japan

Optically active β -(*tert*-hydroxy)esters with good enantiomeric excesses (up to 75% e.e.) are synthesised via enantioselective Reformatsky reaction with ketones in the presence of chiral *N,N*-dialkylnorephedrine.

The enantioselectivities of the addition of metal enolates to prochiral ketones are usually lower than those to prochiral aldehydes.¹ This tendency can be attributed to the greater difficulty in differentiating between the two carbon-containing substituents in ketones than in differentiating one carbon-containing substituent from hydrogen in aldehydes.

Chiral β -(*tert*-hydroxy)esters **4** are important units which are involved in some naturally occurring compounds such as (*R*)-mevalonolactone² and in an intermediate in a synthesis of colchicine.³ Although some diastereoselective syntheses of β -(*tert*-hydroxy)esters have been reported,⁴ enantioselective synthesis of **4** by the Reformatsky reaction⁵ with ketones

using chiral ligands, if it were possible, would become a more direct and convenient method than the diastereoselective methods. To the best of our knowledge, no good method has been reported for the enantioselective Reformatsky reaction with ketones leading to optically active **4**. β -(*tert*-Hydroxy)-esters **4** with only 7–39% enantiomeric excess (e.e.) have been reported in the enantioselective Reformatsky reaction with ketones in the presence of (–)-sparteine.^{6a} Thus, the enantioselective synthesis of **4** with higher e.e.s is a challenging problem. We have reported the enantioselective alkylation of aldehydes (and not ketones) with dialkylzinc reagents using *N,N*-dialkylnorephedrines as chiral catalysts.⁷

Table 1 Enantioselective Reformatsky reaction with ketones

Entry ^a	Ketone 1	Chiral ligand 3	<i>T</i> /°C	<i>t</i> /h	β -(<i>tert</i> -Hydroxy)ester 4			
					Yield (%)	E.e. (%) ^b	Config. ^c	
1	1a	(1 <i>S</i> ,2 <i>R</i>)- 3a	–13	25	4a ^d	65	74	<i>S</i>
2	1a	(1 <i>R</i> ,2 <i>S</i>)- 3a	0	21	4a	47	74	<i>R</i>
3	1a	(1 <i>R</i> ,2 <i>S</i>)- 3a	0	21	4a	21	73	<i>R</i>
4	1b	(1 <i>S</i> ,2 <i>R</i>)- 3a	–13	45	4b ^e	38	75	
5	1c	(1 <i>S</i> ,2 <i>R</i>)- 3a	–13	24	4c ^f	57	73	
6	1a	(1 <i>S</i> ,2 <i>R</i>)- 3b	0	18	4a	35	65	<i>S</i>
7	1a	(1 <i>S</i> ,2 <i>R</i>)- 3c	0	23	4a	17	61	<i>S</i>
8	1a	(1 <i>S</i> ,2 <i>R</i>)- 3d	0	21	4a	36	50	<i>S</i>

^a Solvent was THF–toluene, 5:3 (v/v). Molar ratio for entries 3, 6, 7 and 8, **1**:**3**:**2** = 1:1:5. For other entries, **1**:**3**:**2** = 1:2:10. ^b Determined by HPLC analyses using a chiral column (0.46 × 25 cm). Eluent: 1% propan-2-ol in hexane; column temperature 40 °C; 254 nm UV detector. For (*S*)-**4a**, chiral column: Chiralcel OD; flow rate: 0.3 ml min^{–1}; retention time (*t*/min): 17.29 (*S*)-minor, 18.78 (*R*)-major. For **4b**, chiral column: Chiralpak AD; flow rate 1.0 ml min^{–1}; retention time (*t*/min): 14.19 minor, 17.87 major. For **4c**, chiral column: Chiralcel OD; flow rate: 0.3 ml min^{–1}; retention time (*t*/min): 17.63 minor, 19.49 major. ^c Configuration was determined as follows: reduction of (+)-**4a** (41% e.e.) [α]_D²⁴ + 4.41° (c 4.63, C₆H₆) with lithium aluminium hydride afforded (*S*)-(–)-3-phenylbutane-1,3-diol **5** [α]_D²⁴ –11.30 (c 3.67, EtOH) in 44% yield. Literature value for (*S*)-**5**, [α]_D²⁵ –32.2 (c 10, EtOH) (S. L. Abidi and J. L. Wolfhagen, *J. Org. Chem.*, 1979, **44**, 433). ^d [α]_D²⁸ + 8.20 (c 3.1, C₆H₆). ^e [α]_D²⁵ –8.86 (c 2.3, MeOH). ^f [α]_D²⁴ + 9.57 (c 1.9, C₆H₆).

